

Highly Efficient and Low Cost Zinc Electrolyte Purification Using ZINCEXTM Solvent Extraction

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Abstract

In hydrometallurgical zinc production, the purity of loaded electrolyte is critical in order to assure electrolytic SHG zinc quality and at the same time yielding high current efficiency. Metallic impurities adversely affect the zinc electrodeposition process and exert a negative influence on the purity and the morphology of the zinc cathodes. Besides, some impurities decrease the current efficiency due to hydrogen evolution and, as a consequence, the unit energy consumption (kWh/kg Zn) is increased.

In the conventional Roasting, Leaching and Electrowinning (RLE) zinc production process, purification of the solution is achieved by adding zinc dust. Zinc displaces elements below it in the electrochemical series and removes the impurities that have a negative effect on the electrowinning process. Conventional purification techniques using zinc dust are complex and labour intensive, usually requiring large additions of dust (frequently ranging from 5 to 15 kg of zinc dust per cubic metre of solution) and other specific reagents in several unit operations. Careful and extensive analytical control is also compulsory. In spite of the above, there are many harmful impurities such as Mg, Ca, Halogens, Na, K, etc. that cannot be removed at all by conventional purification processes and can significantly impair electrowinning performance.

The ZINCEXTM technology owned by Tecnicas Reunidas represents a simple, highly efficient, and low cost purification process which removes impurities from zinc electrolytes. This commercially available solvent extraction system is a perfect barrier to all metallic impurities, and in addition, Mg, Ca, Halogens, Na, K, etc, are also rejected. As the ZINCEXTM solvent extraction process does not consume zinc dust, the electrolytic zinc production and the income of the zinc refinery is 5-10% higher in comparison to the RLE technology; as a result, this innovative technology presents around 15% lower operating cost and similar capital cost compared with the conventional purification process.



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Proprietary Technology Development Division TECNICAS REUNIDAS, S.A., Spain



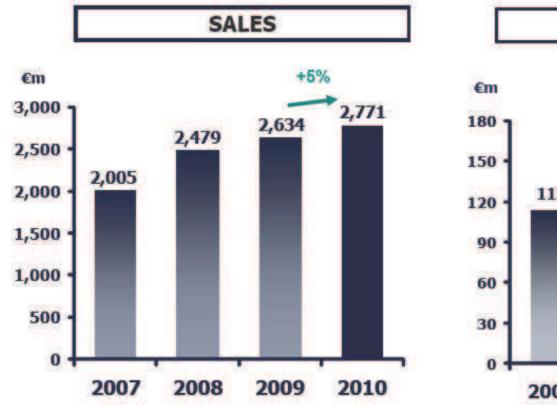


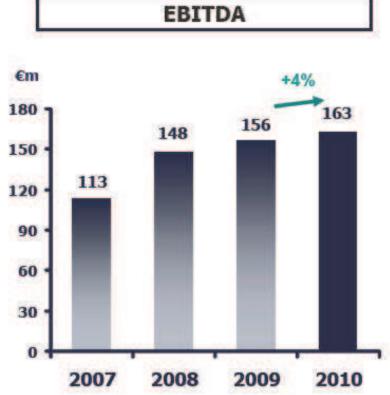
TECNICAS REUNIDAS

PROPRIETARY TECHNOLOGIES DEVELOPMENT DIVISION



COMPANY BACKGROUND





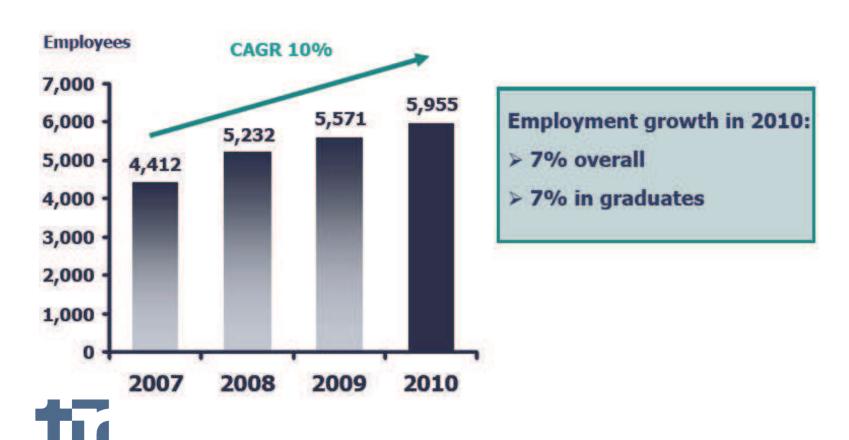


COMPANY BACKGROUND

⇒ 50 Years Experience; More than 1000 Projects worldwide

Headquarters Spain + TR Abroad

TECNICAS REUNIDAS



BUSINESS AREAS

TR Group

OIL & GAS

DOWNSTREAM

- Refining
- Petrochemical Know-how
- Refining Units

Proprietary know-how:

Heat transfer

UPSTREAM & NATURAL GAS

- Oil & Gas
 Development
 Fields (Onshore/
 Offshore)
- LNG, Liquefaction, Regasification & Storage
- Offshore & Pipelines

POWER

- Combined Cycles
- Nuclear
- Conventional thermal Plants
- Renewable energy
- Cogeneration

INFRA-STRUCTURES

- Water treatment
- Desalination
- Waste management
- Air transport
- Land transport
- Ports

PROPRIETARY TECHNOLOGY DEVELOPMENT

Espindesa:

- Nitric Acid
- Amonium Nitrate
- Fertilizers

<u>Hydrometallurgy &</u> <u>Electrochemistry:</u>

- Base Metals: Zinc, Lead, Copper,...
- Precious metals,
- Energy Storage
- Recycling



OVERVIEW:

- > INTRODUCTION
- > CONVENTIONAL ZINC DUST PURIFICATION
- > ZINCEXTM SOLVENT EXTRACTION PURIFICATION
- > COMPARISON OF THE PURIFICATION TECHNOLOGIES
- > INDUSTRIAL CASE STUDY
- **CONCLUSIONS**



INTRODUCTION

- In hydrometallurgical zinc production, the purity of loaded electrolyte is critical in order to assure:
 - Electrolytic SHG zinc quality
 - Yielding high current efficiency
- Impurities adversely affect the zinc electrodeposition:
 - Negative influence on cathode purity
 - Bad deposit morphology: dendrites, sticky, etc.
 - Current efficiency decreasing: H₂ evolving
- In consequence:

Loaded electrolyte purification is likely the most important and critical unit process in any zinc refinery



INTRODUCTION

- In the conventional Roasting, Leaching and Electrowinning (RLE) zinc production process, purification of the pregnant solution is achieved by adding zinc dust:
 - Complex and labour intensive operation where careful and detailed analytical control is compulsory
 - Large addition of dust: 5-15 kg Zn dust/m³ solution
- In spite of the above, there are many harmful impurities that are not removed at all by zinc dust purification:
 - Mg, Ca, Halogens, Na, K, etc

Such impurities can impair electrowinning performance



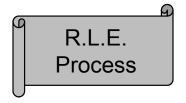
INTRODUCTION

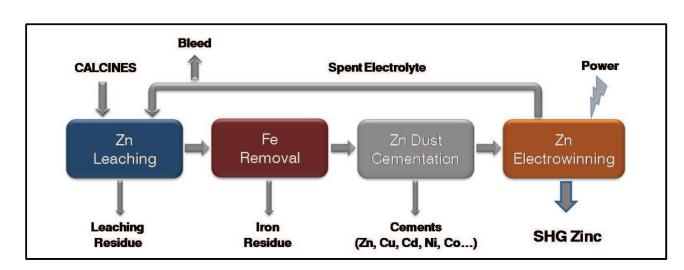
- The ZINCEXTM solvent extraction technology owned by Tecnicas Reunidas is a highly efficient and low cost purification process for zinc electrolytes and represents an ideal solution versus conventional purification:
 - ➢ Given that zinc dust is not required, the income of the zinc refinery increases by 5-10% because all zinc plates is sold, none is recycled as zinc powder
 - Solvent extraction is totally automatic, requiring low labour and maintenance: low-cost process
 - Solvent extraction system is a perfect barrier to all metallic impurities, and in addition, Mg, Ca, Halogens, Na, K, etc, are fully rejected

This presentation compares the conventional purification versus the new ZINCEXTM solvent extraction purification from technical and economic viewpoints

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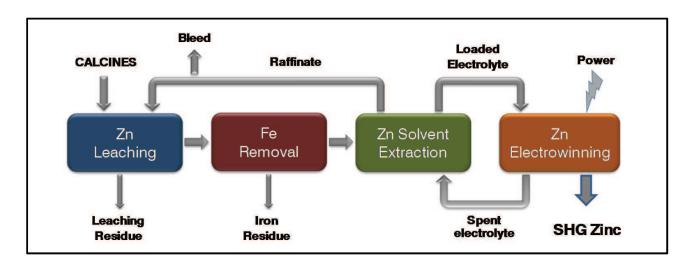
CONCEPTUAL BLOCK DIAGRAMS











CONVENTIONAL ZINC DUST PURIFICATION

• Electrolyte purification is achieved by means of zinc dust cementation:

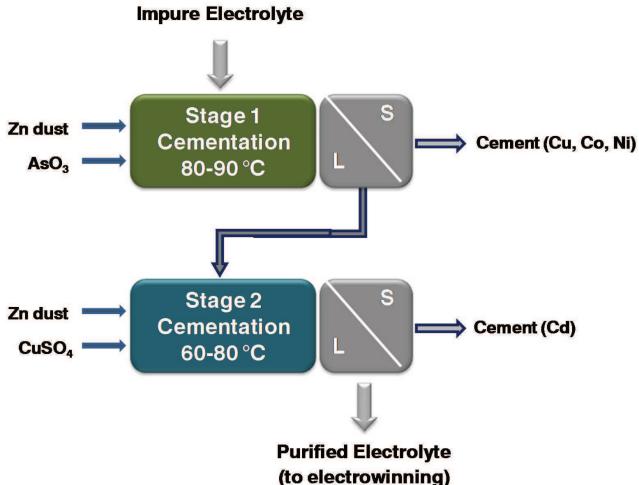
$$Me^{n+} + n/2 Zn^0 \rightarrow n/2 Zn^{2+} + Me$$

("Me"= Cu, Cd, Ni, Co, Ge, etc)

- Achieving an extremely low level of impurities is a complex process that requires high excess of zinc dust and also specific additives to catalyse the reactions
- > Several mechanisms are proposed, e.g. formation of intermetallic compounds that promotes impurities deposition
- The cementation procedures are tailored and adapted to every zinc refinery depending on the composition and characteristics of the impure solution, the capacity, batch or continuous process, etc
- Most applied methods are summarised as follows:

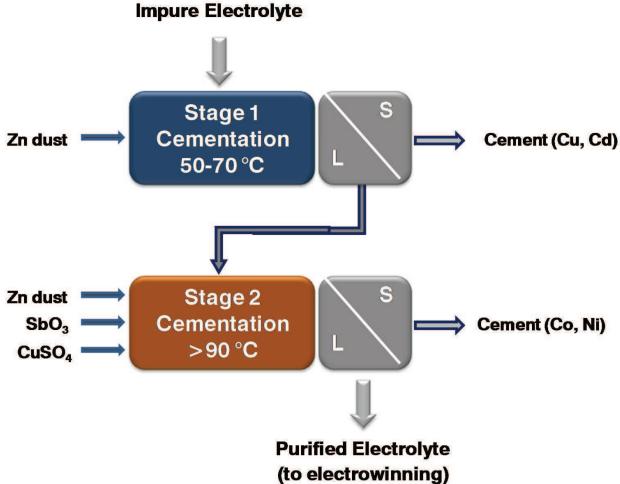


-Two-stage "hot-cold" arsenic trioxide process-



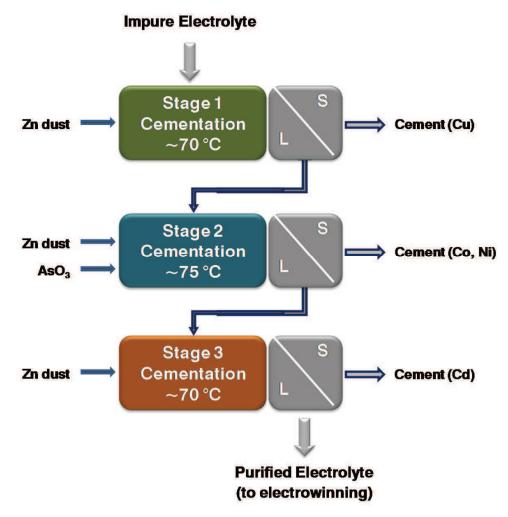


-Two-stage "cold-hot" antimony trioxide process-





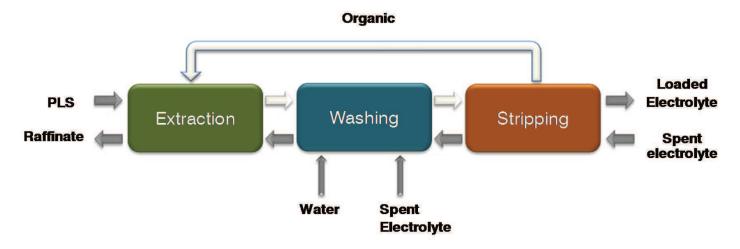
-Three stage process-





ZINCEXTM SOLVENT EXTRACTION PURIFICATION

The solvent extraction process consists of three main stages: Extraction, Washing, and Stripping running in counter-current, which enables production of an ultrapure loaded electrolyte that guarantees SHG zinc quality production:



The organic phase contains di-2-ethil-hexil phosphoric acid (D2EHPA) diluted in kerosene, and the operating conditions are adapted to achieve a perfect separation of the zinc metal from the remaining impurities

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ZINCEXTM SOLVENT EXTRACTION

Extraction stage:

2 RH (org) +
$$Zn^{2+}$$
 (aq) $\rightarrow R_2Zn$ (org) + 2 H⁺ (aq)

Washing stage:

The organic phase is contacted with water and (acidic) spent electrolyte in order to remove the entrained or co-extracted impurities

Stripping stage:

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$$R_2$$
Zn (org) + 2 H⁺(aq) \rightarrow 2 RH (org) + Zn²⁺(aq)

This zinc solvent extraction system operates automatically and on a continuous basis: low labour and low energy; it is a simple and cost-efficient process

PROCESSES COMPARISON: Reagents

| Zinc Dust Cementation | ZINCEX [™] Solvent Extraction |
|---|--|
| 5-10% of zinc production is converted to zinc | Zinc dust is not required |
| dust | |
| Yearly income is 5-10% reduced due to zinc | Yearly income increased by 5-10% because |
| dust production | zinc dust is not required and all produced |
| | electrolytic Zn is sold |
| 5-10% overdesign of the tankhouse due to | Capex of the tankhouse is 5-10% lower as |
| zinc dust production | zinc dust is not required |
| Reagents include zinc dust, arsenic or | The only reagent required is the organic |
| antimony trioxide, copper sulphate, etc | make up |
| Formation of H ₂ gas due to reaction of zinc | No formation of H ₂ or AsH ₃ gas |
| dust and acid, and possible formation of AsH ₃ | |
| Organic compounds are not used | Risk of fire due to organic compounds. A safe |
| | design is required to avoid any trouble |
| Organic compounds are not used | Risk of traces of organic material passing to |
| | the tankhouse. This is avoided through |
| | proper design and operation of active carbon |
| | columns |



PROCESSES COMPARISON: Labour & Control

| Zinc Dust Cementation | ZINCEX [™] Solvent Extraction |
|---|---|
| Labour is required for: strict analytical | Automatic and continuous process with |
| control, filtration and handling of cements, | minimum analytical control; as a |
| preparation and dosing of additives, | consequence, little labour is required |
| sometimes batch purification operations, etc | |
| The process sometimes fails to meet the | The solvent extraction process is based on |
| target level of impurities in the electrolyte | equilibrium, therefore impurity levels in the |
| | electrolyte are always kept constant |
| Re-dissolution of impurities from zinc | No risk of re-dissolution of impurities becasue |
| cements when the retention time is long | zinc cements are not produced |



PROCESSES COMPARISON: Temperature

| Zinc Dust Cementation | ZINCEX [™] Solvent Extraction |
|---|---|
| Electrolyte is heated to 90-95 °C in order to | Electrolyte temperature is kept constant at |
| achieve efficient purification | 40-45 °C |
| High solubility of gypsum due to high | The loaded electrolyte is gypsum free |
| temperature, creating scaling and | because this solvent extraction system runs |
| crystallisation problems in the cellhouse | as a barrier to calcium |



PROCESSES COMPARISON: Purification Effic.

| Zinc Dust Cementation | ZINCEX [™] Solvent Extraction |
|---|--|
| | |
| Purification is carried out on the main | Purification is carried out on the effluent |
| process stream, so any problems in the | bleed, so any operating problems will never |
| purification process may affect tankhouse | affect the tankhouse performance |
| operation | |
| Single circuit, thus any perturbation of the | Independent solvent extraction and |
| purification stage may lead to Fe, Ni, As, etc, | electrowining circuits, thus surges ofh as Fe, |
| being fed to the tankhouse with very | Ni, As, etc, are buffered by the organic |
| detrimental consequences | phase, and in consequence, the loaded |
| | electrolyte quality remains unchanged |
| Efficient process for removal of metallic | This solvent extraction system is a perfect |
| impurities, but unable to reject other ions | barrier for metallic impurities and other |
| such as Mg, Ca, Cl, F, Mn, Na, K, etc | elements such as Mg, Ca, Mn, Na, K, Cl, F, |
| | etc, which are fully rejected |



PROCESSES COMPARISON: Zinc Losses

| Zinc Dust Cementation | ZINCEX [™] Solvent Extraction | |
|--|--|--|
| The copper cements is contaminated with As | Cements are not produced | |
| or Sb arising from purification additives | | |
| The complex composition of the cements | Cements are not produced | |
| makes processing them difficult | | |
| Losses of zinc content in zinc cements | Cements are not produced, losses of zinc are | |
| | therefore minimal | |



PROCESSES COMPARISON: Effect on EW

| Zinc Dust Cementation | ZINCEX [™] Solvent Extraction |
|---|--|
| Ultrapure electrolyte production is not | Produces ultrapure loaded electrolyte |
| possible because some elements such as | presents the following advantages: (i) Lower |
| Mg, Ca, Mn, Na, K, Cl, F, etc, are not | cell voltage and power consumption |
| rejected | (alkalines), (ii) Longer electrode life (Cl, F), |
| | (iii) Minimisation of cell sludge (Ca, Mn), (iv) |
| | Highest zinc quality, (v) Lower anode |
| | cleaning frequency (Ca, Mn) |
| Loaded electrolyte produced is saturated with | Produces ultrapure loaded electrolyte, free of |
| gypsum | gypsum, presents the following advantages: |
| | (i) No gypsum scaling problems, (ii) Simpler |
| | and lower cost cooling system, (iii) Minimises |
| | maintenance requirements in the cooling |
| | towers |



PROCESSES COMPARISON: Process Stages

| Zinc Dust Cementation | ZINCEX [™] Solvent Extraction |
|--|---|
| Cu, Cd Removal | |
| CI, F Removal | |
| Ni, Co Removal | |
| Gypsum Removal | Solvent Extraction (This single process stage is equivalent |
| Magnesium Removal | to 8 process stages required in a conventional zinc refinery) |
| Manganese Removal | |
| Zinc Dust Production Unit | |
| Heating and Cooling Electrolyte System | |



INDUSTRIAL CASE STUDY

A comparative techno-economic study has been undertaken for a zinc refinery having a capacity of 100,000 t/a SHG zinc:

OPEX (reagents and consumables, manpower, maintenance and 10% contingency)

| Operating Cost | Zinc Dust Cementation | ZINCEX [™] Solvent Extraction |
|----------------------|-----------------------|---|
| TOTAL, USD/t SHG Zn= | 105 | 58 |

CAPEX

| Investment Cost | Zinc Dust Cementation | ZINCEX TM |
|---------------------|----------------------------|----------------------|
| (license excluded) | (including gypsum removal) | Solvent Extraction |
| TOTAL, Million USD= | 50 | 52 |



CONCLUSIONS (1)

- Zinc electrolyte purification using ZINCEX[™] solvent extraction technology presents many advantages in comparison with conventional zinc dust purification process, such as:
 - Better overall process performance
 - Enhanced electrowinning
 - Lower operating cost (approximately 40% saving per ton of Zn)
 - Yearly income increased by 5-10% because zinc dust addition is not required, and therefore, all produced electrolytic zinc is sold
 - Ability to treat difficult materials, e.g. containing high Mn, Cl, F, Mg, etc



CONCLUSIONS (2)

The positive economic results obtained in this preliminary study confirm that ZINCEXTM solvent extraction is a reliable and cost-efficient alternative able to substitute the conventional zinc dust purification in RLE zinc refineries



